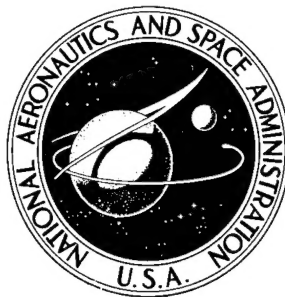
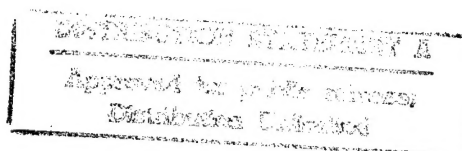


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# ADHESION OF POLYMERS TO TUNGSTEN AS STUDIED BY FIELD ION MICROSCOPY

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16. Abstract <p>Mechanical contacts with polytetrafluoroethylene (PTFE) and polyimide polymer (PI) contacting tungsten field ion microscope emitter tips were conducted in vacuum. The metal polymer interface was examined on an atomic scale by helium field ion microscopy. Mechanical transfer was observed upon simple touch contact of the polymer materials to tungsten. The transferred polymers were stable even at the high field necessary for helium ionization, thus implying a strong chemical bonding to the metal surface. The mechanical contact stresses orient the transferred material on the tungsten surface in a radial manner from the center of contact. With sufficient load the PTFE and PI cause deformation of the tungsten. On the tungsten (110) plane, individual molecular clusters can be resolved.</p>			
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# ADHESION OF POLYMERS TO TUNGSTEN AS STUDIED BY FIELD ION MICROSCOPY

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Lewis Research Center

## SUMMARY

Organic polymers are used extensively in lubrication systems. Fundamental to using these materials as lubricants is adherence between the polymers and metal surfaces. In this investigation, experiments were conducted using the field ion microscope (FIM) to study and characterize the polymer-metal interface on an atomic scale. Mechanical contact between the polymeric material polytetrafluoroethylene (PTFE) and polyimide polymer (PI) with the clean tungsten field ion emitter tip were made in situ in the FIM. Both the PTFE and PI transferred to the tungsten metal tip on simple mechanical touch contact. Furthermore, the images of the transferred material were stable with helium ion imaging. Thus, the transferred polymers were very strongly bonded which implies chemical bonding (i. e. , the formation of an organometallic bond).

Furthermore, the transferred polymeric fragments were oriented on the tungsten surface in a manner suggesting they were aligned by the contact stress radially outward from the center of the contact zone. With heavier loads both the PTFE and PI caused deformation of the tungsten tip. On the (110) plane, certain individual polymer molecular clusters could be resolved, in particular. Some PTFE fragments are observable which resemble the PTFE structure and which may represent a single monomer of the polymer chain.

## INTRODUCTION

In a lubrication system, the relative motion of the components gives rise to friction and wear. The friction and wear properties of materials in sliding contact can be related to the behavior of the materials at the sliding interface. With metals in sliding contact, adhesion at the interface is the significant factor in the friction and wear process. To reduce the friction and wear a lubricant is used. For many applications conventional

lubricants such as oils and greases cannot be used. In these cases solid lubricant materials, both inorganic and organic, often are used.

A common organic solid lubricant material is polytetrafluoroethylene (PTFE). PTFE has been studied under vacuum conditions with sliding friction experiments and as a component in bearing systems (refs. 1 and 2). The very stable polyimide polymer (PI) materials have also been studied (ref. 3) and are employed as cage materials for bearings.

Fundamental to using these materials is their adherence to the sliding surface. The adherence of the material during application and subsequently during running determines to a large extent its ability to function as a lubricant. Thus, a fundamental understanding of the adherence of a solid lubricant to metal surfaces will significantly aid in the selection and application of potential lubricant materials.

Previous work has shown that the field ion microscope is a powerful tool for fundamental studies of the adhesion and surface characterization of metal-metal contacts (refs. 4 and 5). The purpose of this investigation was to determine if the field ion microscope could be used to study the polymer-metal interface and, if so, to characterize the interaction of the lubricant with the metal surface.

To accomplish this objective experiments were conducted in which mechanical contacts were made with solid lubricant materials contacting the field ion microscope emitter tip. The contacts were made in situ in the field ion microscope vacuum system. The solid lubricant materials examined in contact with tungsten on the metal emitter tip were polytetrafluoroethylene (PTFE) and polyimide polymer (PI).

## APPARATUS

### Field Ion Microscope (FIM)

The field ion microscope is shown schematically in figure 1. The microscope is of metal construction with copper sealing gaskets so the system can be baked out to achieve low background pressures. The field emitter is mounted by tack welding to a nickel wire loop which slips over two copper posts extending down from the end of the cryogenically cooled cold finger. Also attached to the copper posts is the high voltage (0 to 30 kV) lead. The copper post mounting disk is separated from the metal (ground) end of the cold finger by sapphire disks which serve both as electrical insulators and thermal conductors. The cold finger is cooled by either filling it with liquid nitrogen (78 K), liquid helium (4.2 K), or by using a two-stage commercial cryostat which can be inserted down the cold finger tube to liquify hydrogen (21 K). The cold finger is welded to a flange which mates with a bellows assembly used to vary the vertical position of the tip. Tip to screen distance can be varied from 8 to 5 centimeters by bellows compression.



The screen of the FIM is a tin oxide coated fiber optic window 7.6 centimeters in diameter with an average fiber diameter of 16 microns. Phosphor is coated on the inside of the fiber optic screen by using an artist's air brush. Photography of the field ion image is done by direct contact photography methods. The film is pressed against the outside of the fiber optic window, and this essentially uses the fiber bundles to transmit the image from inside the screen to the emulsion.

The image gas system consisted simply of cylinders of research grade helium and neon which were bled into the evacuated microscope to a pressure around 3 millitorr with a variable leak valve.

The vacuum system of the field ion microscope consisted of a cryosorption fore pump, a 20/400 liter per second combination ion pump and water cooled titanium getter pump, and a noble gas pumping 50 liter per second ion pump. Chamber pressure was monitored at imaging pressure by a thermocouple gage and at UHV conditions with a triggered discharge gage. With moderate bakeout ( $250^{\circ}\text{C}$ ), chamber pressure was  $10^{-10}$  torr.

The electrical schematic of the FIM is shown in figure 2. A regulated, reversible, 0 to 30 kilovolt dc power supply is used in the positive mode for field ionization. A 2-nanosecond pulser with variable pulse height from 0 to 2 kilovolts is floated at the 30-kilovolt supply bias level. The pulser is capable of single manual pulsing or repetitive pulsing at 60 cycles per second. The pulser permits controlled field evaporation of the specimen at rates from 1 atomic layer to several layers per second. Between the HV supply and the pulser is a 500-megohm resistor for preventing damaging electrical breakdowns inside the microscope. Also floated in series is a solid state electrometer for measuring the ion and emission current. The electrometer is protected against breakdown by an input to chassis diode couple (27-v breakdown voltage).

### Emitter Contactor Apparatus

The manipulator which was used for contacting the field ion emitter tip is shown schematically in figure 3. Basically the device is a stainless steel beam mounted on a taut band. On one end of the beam is a small permanent bar magnet. Above and below the bar magnet, outside the glass vacuum tube, are two small electromagnets wired so that like poles face the permanent magnet. The beam can be made to move either up or down by varying current to the electromagnets. On the other end of the beam is a platform which by the compression of the horizontal bellows, is moved under the field emitter tip for contact. On the platform is the material with which the tip is touched.

Also mounted on the beam is a stainless steel flag which acts to intercept the light from a small lamp to a photocell. The shadow cast on the photocell changes the resistance of the cell, which causes a change in voltage to the electromagnets. This

photocell-electromagnet voltage system serves to damp out vibrations. The photocell voltage also serves as a monitor for beam position. Photocell voltage is amplified and read on a multirange recorder. Any change in beam position is reflected in photocell voltage which is recorded.

The contact of the tip was sensed by using the deflection of a thin spring. In this case a foil of platinum was welded to a linear spring. When contact with the field tip occurs, the spring deflects enough so the foil makes an electrical contact with an electrode just below the foil. The spring and foil are grounded, thus causing a measurable current between the electrode and ground which is read on a meter outside the microscope. The gap between the foil and the electrode can be set to vary the force required to give a current indication. The gap spacing could be set close enough so that loads less than 0.25 milligram could be sensed. Even this small load is sufficient to produce a high contact stress.

Precalibration of contact force as a function of electromagnet current is accomplished by using an electronic balance with a sensitivity of 0.01 milligram. A known load can be applied to the contacted tip by applying more electromagnet current to the top magnet. After contacting and loading is completed, the current to the top electromagnet is reduced until the tip and contact material separate.

To reduce the influence of ambient vibrations on the experiments, the entire field ion microscope including pumps, power supplies, gages, gas supply systems, etc., and the emitter contactor apparatus are mounted on a vibration isolation table. The vibration table uses air piston suspension and has a natural frequency of 1.1 hertz with better than 90 percent isolation at resonant frequencies of 7 to 20 hertz.

## PROCEDURE

The tip contact procedure is shown as a flow diagram in figure 4. The tip is electro-etched from fine high purity wire to an extremely fine radius ( $<500 \text{ \AA}$ ). Methods for tip preparation are discussed in the literature (ref. 6). Following etching, the specimen is mounted on the cold finger and the FIM is pumped into the UHV range.

The tip is imaged at a pressure around 3 millitorr with either helium or neon gas. Field evaporation is conducted to increase tip radius and to prepare a nearly perfect crystallographic surface. After field evaporation, the image gas is pumped out while the high voltage is kept on to prevent contamination of the tip. After pressure is less than  $10^{-9}$  torr, the high voltage is turned off. A clean surface can be maintained for several minutes at pressures below  $10^{-9}$  torr.

Following the pump down, the emitter contact apparatus is moved under the tip by bellows compression. The platform is then raised until the tip contacts the solid lubricant material as sensed by the electrode contacting the metallized underside of the

lubricant material. At this point additional loading may be applied beyond that required to close the preset gap (0.25 mg). The load is then removed and the image of the surface reestablished by readmitting the image gas and turning on the high voltage. The stereographic projection of the body centered cubic lattice is shown in figure 5.

## RESULTS

### Polytetrafluoroethylene - Tungsten

Mechanical contacts of tungsten emitter tips by PTFE were made in vacuum of  $10^{-10}$  torr. Figure 6 shows the precontact image of a somewhat imperfect tip. The dark area is a furrow between two protruding asperities. Figure 7 is the same tip after being contacted with PTFE. The surface is characterized by a random appearance containing many bright spots. Note the diffuse tails extending outward from many of the spots. These diffuse tails are more evident in figure 8 which was taken at a slightly higher voltage. There are apparent molecular structures which can be resolved on the (110) plane. It is likely that many of the other bright spots are more than one atom but cannot be resolved because of their enhanced brightness. These apparent molecular structures have an appearance very similar to the end view of a PTFE chain. Structural illustrations of the PTFE chain and of the PI structure are shown in figure 9. It is thus believed that some PTFE has transferred to the tungsten surface and that the molecular images resolved on (110) plane represent a fragment of a PTFE chain, possibly a single monomer. Furthermore, the bonding is strong enough to withstand the helium image field without desorbing. This strong bonding suggests a tungsten-carbon bond formed at the interface. Figure 10 is a micrograph of the same surface taken at 18.0 kilovolts. Several of the bright spots that appeared in figures 7 and 8 have now desorbed. In addition, the remaining ones do not exhibit the diffuse tails that were observed previously. It is not clear what causes these tails. During metal-metal contact, reference 5, there was no evidence of diffuse tails as was observed with the PTFE. It is of interest to note that the tails appear to be projected outward from a point near the bottom of the micrograph. This point may represent the center of the mechanical contact and the tail-like structure may be due to an orientation effect caused by the stresses exerted radially from the center of the contact.

Figure 11 is a micrograph taken at 17.5 kilovolts after raising the voltage to 19.0 kilovolts. Some further desorption has occurred, but generally the image is similar to that of figure 10. The micrograph in figure 12 was taken after raising the voltage to 22.0 kilovolts in order to desorb the PTFE from the (110) plane.

Repeating the experiments with lightly loaded tungsten-PTFE contacts showed similar results.

A heavily loaded tungsten-PTFE (approximately three times more load) contact gave the rather surprising result that extensive deformation of the tungsten occurred. Figure 13 shows the precontact micrograph, and figure 14 shows a postcontact micrograph. This photograph was taken after the desorption of the adhered PTFE. The deformation extended far into the bulk of the material as shown in figure 15, which was taken after extensive field evaporation (an effective increase in tip radius of approximately fifty percent). Generally, during heavy loaded contacts the tip will bend so that the image is often displaced on the screen; however, there is no displacement of (110) in these micrographs. The tip, when pressed into the PTFE, may be prevented from bending and thus absorbing the stress further up the shank by the very large area of contact (relative to metal-metal contacts).

## Polyimide - Tungsten

Mechanical contacts with a polyimide polymer (PI) contacting tungsten tips were made in vacuum of  $10^{-9}$  torr with both light and heavy loads. Figure 16 is a precontact micrograph of tungsten. This particular tip had been contacted previously as there is some deformation in the vicinity of the (100) plane. Figure 17 shows a micrograph after the light load PI contact. The surface is again characterized by a random distribution of bright spots. Figure 18 is the same tip with an increase of 500 volts during imaging. Many of the spots in figures 17 and 18 are elongated and appear to contain groups of atoms as if they were transferred molecular fragments of PI. Figure 19 is a micrograph taken at 14.5 kilovolts. Considerable desorption has occurred and the underlying tungsten is visible. Note the surface has not been displaced nor is there any obvious evidence of further deformation. As with PTFE, the (110) plane appears to have several polymer fragments adhered to it. Figure 20, taken at 15.0 kilovolts, shows further desorption has occurred, particularly on (110). It is of interest to note the change in the character of the remaining spots on (110) as the neighboring fragments desorb. Additional increases in voltage cause further desorption; however, it was necessary to evaporate the tungsten substrate to return the surface to the ordered tungsten appearance.

To determine if load would significantly influence the PI-tungsten contacts, an experiment was conducted with PI contacting tungsten using a load approximately three times greater than that used for light load contents. Figure 21 is the precontact micrograph. Figure 22 is the image of the same tungsten tip after a heavily loaded PI contact. The results show a series of elongated spots all radially projected outward from a dark nonimaging area on the right side of the micrograph. As with PTFE, it appears that the polyimide polymer fragments have been oriented by the radial stresses projecting outward from the center of contact, where the center of contact is the dark region to the right in the micrograph. Figure 23 is a micrograph taken with a slightly higher voltage.

It can be seen that the (110) plane has been displaced to the upper left indicating the tip is slightly bent by the force of the contact. Raising the voltage to 10.25 kilovolts caused considerable desorption of the streak to occur; this condition resulted in a surface that had an appearance very similar to the light load PI contact of figure 17.

Subsequent micrographs taken with increased voltage show similar results to those light load PI-tungsten contact micrographs.

## DISCUSSION

The results of this study with PTFE and PI polymers in contact with tungsten indicate that polymers will transfer to a clean metal surface on simple touch contact and that the bond formed reflects more than simple mechanical transfer. The stability of the transferred polymers at the high field for helium ion imaging imply that the bond of the polymer to the metal surface is chemical in nature. With PTFE it is hypothesized that the bonding is that of carbon to the metal surface because the carbon to carbon bond is the weakest bond in the PTFE structure and the one most frequently seen broken on polymer scission (ref. 7). Furthermore, the carbon could readily interact with the clean tungsten to form bonds stable at the imaging and field evaporation voltages applied in this study.

With the polyimide (PI) both oxygen and carbon bonding to the tungsten surface are possible. But again because of the imaging and evaporation voltages employed the bond must be chemical in nature.

The concept of chemical bonding of the polymers to the clean metal surface necessitates the breaking of bonds in the organic molecule and the subsequent formation of metal to carbon or oxygen bonds to form organometallics. The breaking of organic bonds by metal surfaces is observed with hydrocarbons contacting metals in the field of catalysis. The tendency for such reactions should be increased when the metal surface is atomically clean because of the enhanced surface activity of the metal.

The effect of loading in the transfer of the polyimide to the tungsten surface indicates an increased amount of polymer transfer with an increase in load. The polymer appears to remain on the tungsten surface in longer chain fragments indicating that fracture occurred deeper in the polymer body than was observed at light loads. When field evaporation was conducted the polymer chain length adhered to the tungsten could be reduced to that observed with light loads.

It is of interest also to note in figures 22 and 23 that preferred orientation of the polymer chains toward the zone of contact has occurred. This in all probability resulted when the tungsten was pressed into the polymer under load. Adhesion of polymer occurred to the tungsten surface. Polymer bonds were broken in the bulk and the relative tangential motion of the polymer body along the radius of the tungsten tip resulted in texturing.

With heavy loading of the polymers against the tungsten surface, field evaporation of the polymer revealed a plastically deformed tungsten surface. This is believed to be the first experimental evidence for the plastic deformation of a metal surface when pressed into contact with a polymer. Tungsten has the highest cohesive energy and one of the highest elastic moduli of the metallic elements in the periodic table. Therefore, it is reasonable to assume that other metals could also be plastically deformed in their surface layers by contact with these polymers with sufficient loads.

## SUMMARY OF RESULTS

Mechanical contacts of tungsten FIM emitter tips by polytetrafluoroethylene (PTFE) and polyimide polymer (PI) were conducted in the field ion microscope, and the effects of these contacts were examined on an atomic scale by field ion microscopy. The following results were obtained:

1. Both the PTFE and PI transfer to a clean tungsten surface on simple touch contact.
2. The stability of the transferred polymeric material at the high fields employed for helium field ion microscopy imply very strong chemical bonding between the polymer and the metal surface. For PTFE this suggests a tungsten-carbon bond, while for PI the bonding could possibly be either that of tungsten to oxygen or carbon.
3. The transferred polymeric materials are oriented by the mechanical contact stress in a radial manner outward from the centers of the contact zones.
4. With sufficient load, both the PI and PTFE can cause deformation of the tungsten substrate.
5. Individual polymer molecular clusters can be resolved on the (110) plane of tungsten.

Lewis Research Center,  
National Aeronautics and Space Administration,  
Cleveland, Ohio, July 29, 1971,  
114-03.

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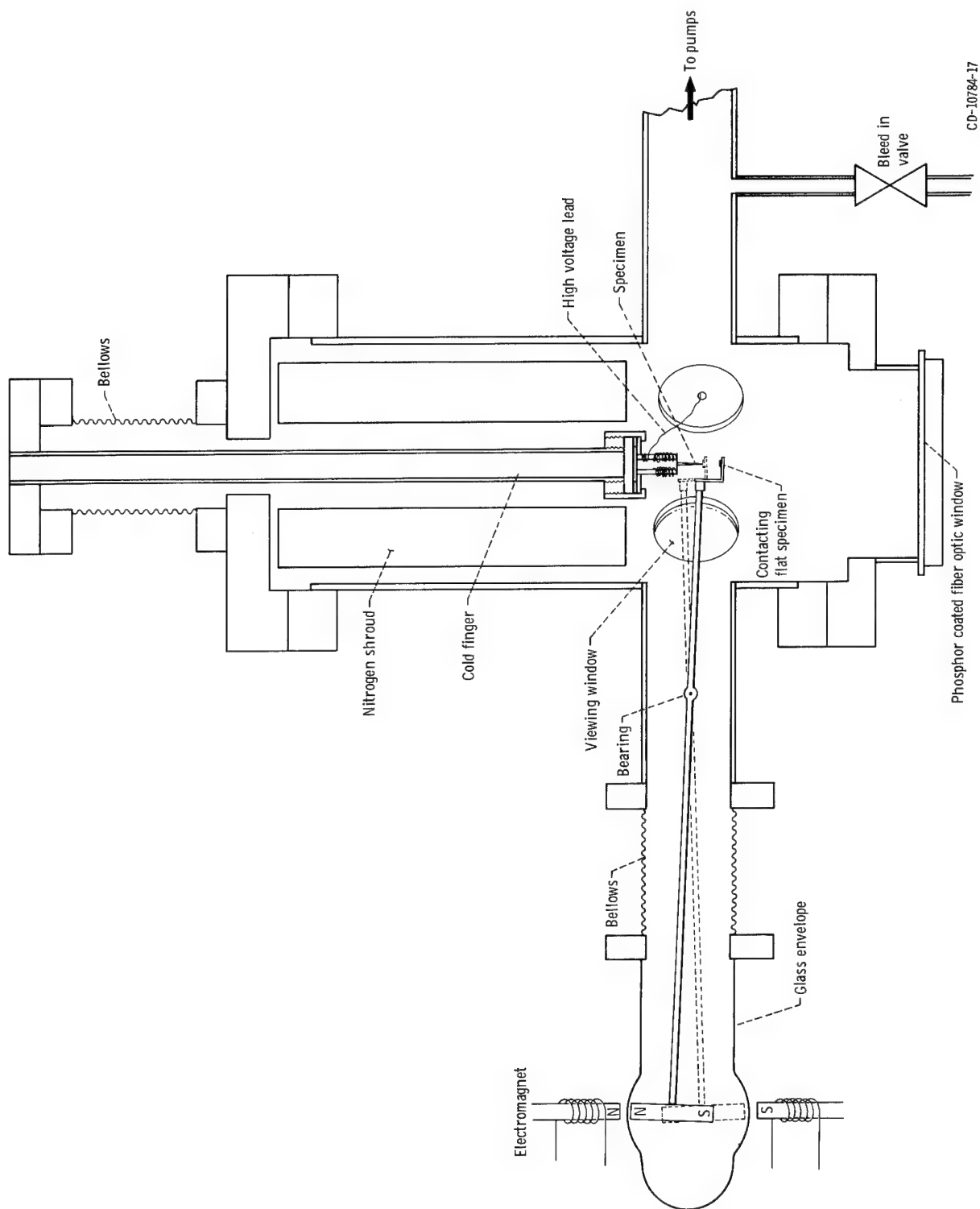


Figure 1. - Field ion emission microscope adhesion apparatus.



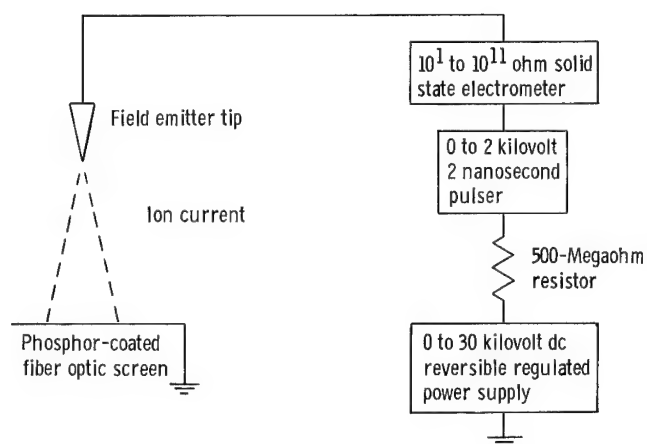


Figure 2. - Electrical schematic for field ion microscope.

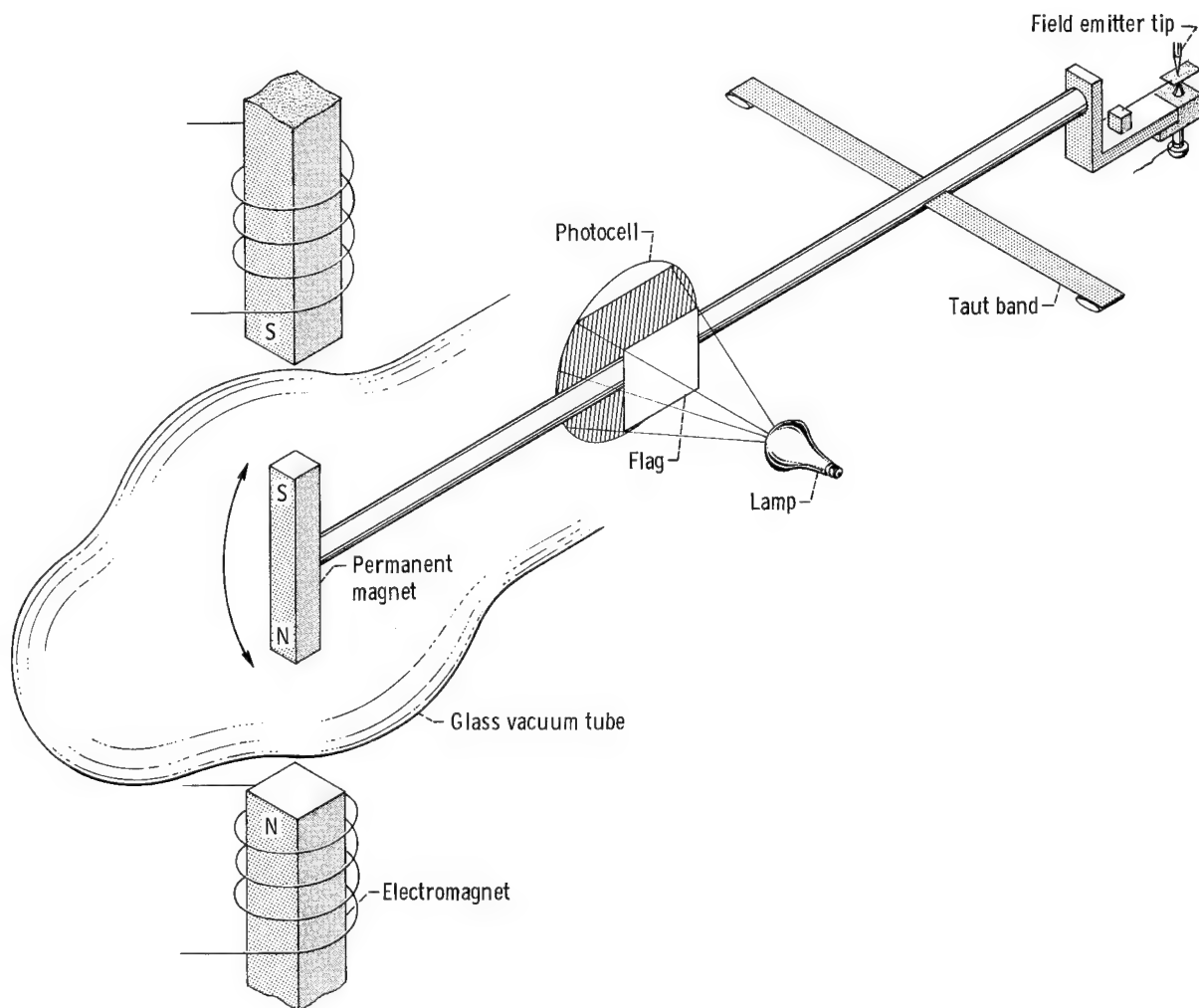


Figure 3. - Field ion microscope emitter tip contact apparatus.

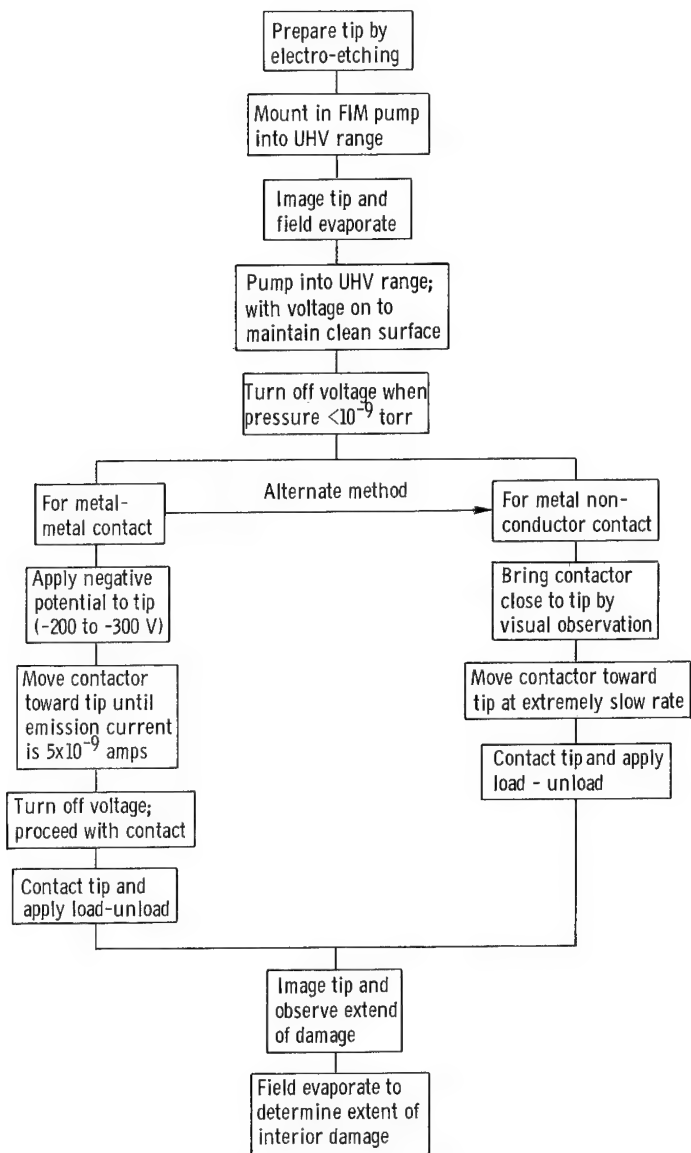


Figure 4. - Flow diagram of FIM tip contact procedure.

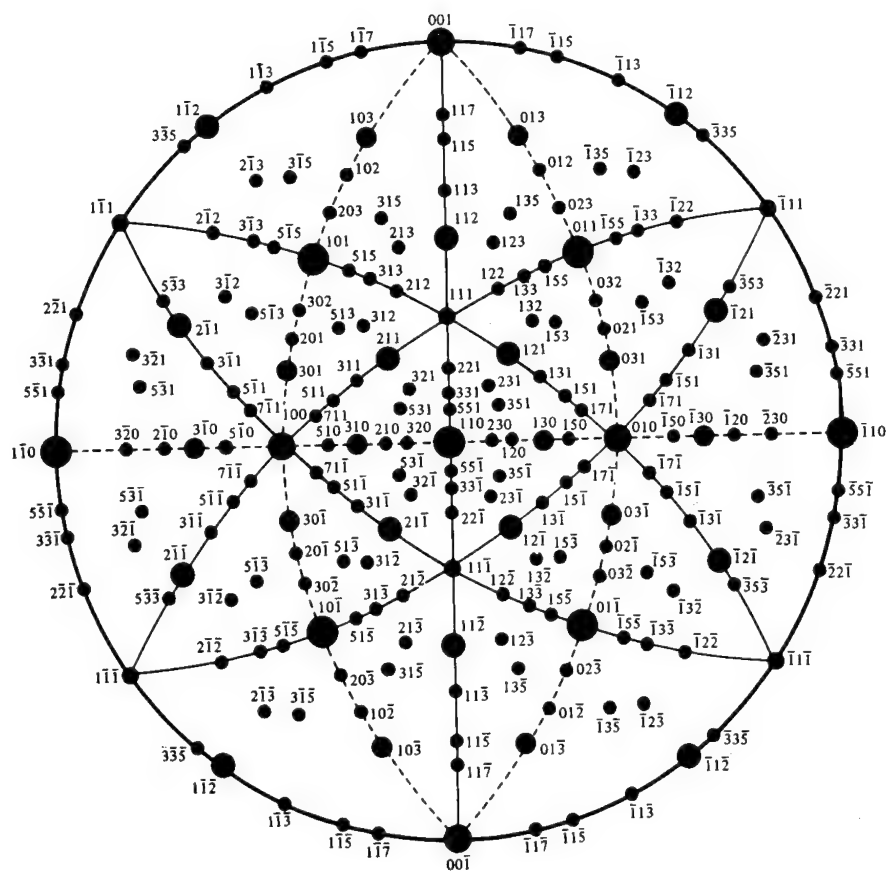


Figure 5. - Stereographic projection of the bcc lattice; (110) oriented.



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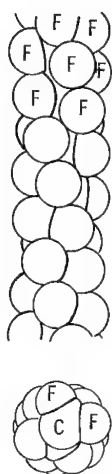
Figure 6. - Tungsten prior to contact (18.0 kV, helium image gas, liquid nitrogen cooling).



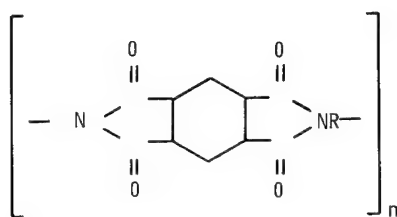
Figure 7. - Tungsten after PTFE contact (16.2 kV, helium image gas, liquid helium cooling).



Figure 8. - Tungsten after PTFE contact (16.5 kV, helium image gas, liquid helium cooling).



Molecular structure of PTFE



Polyimide structure

Figure 9. - Structure of PTFE and PIP.



Figure 10. - Tungsten after PTFE contact (18.0 kV, helium image gas, liquid helium cooling).



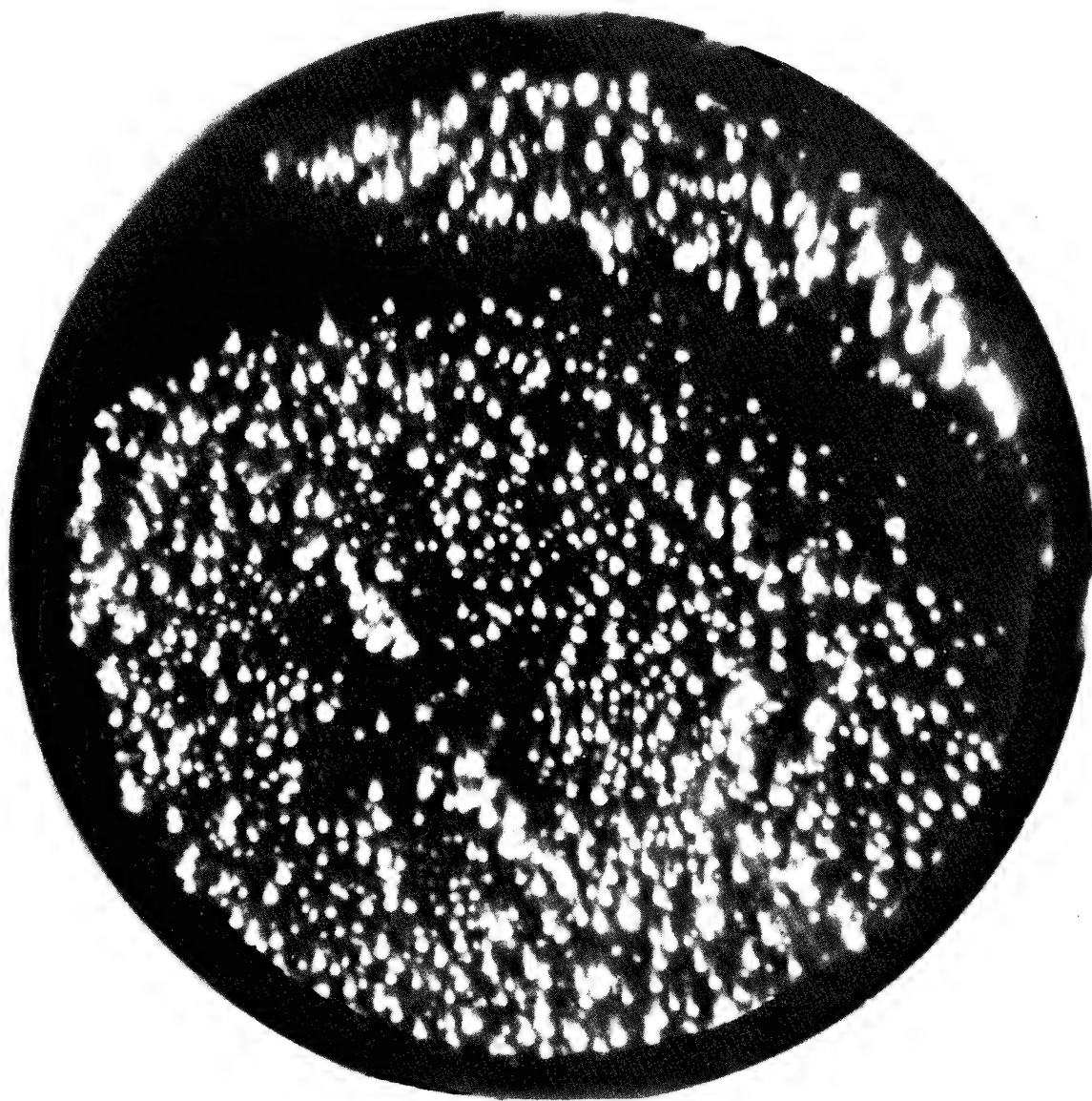


Figure 11. - Tungsten after PTFE contact (17.5 kV, helium image gas, taken after raising voltage to 19.0 kV, liquid helium cooling).

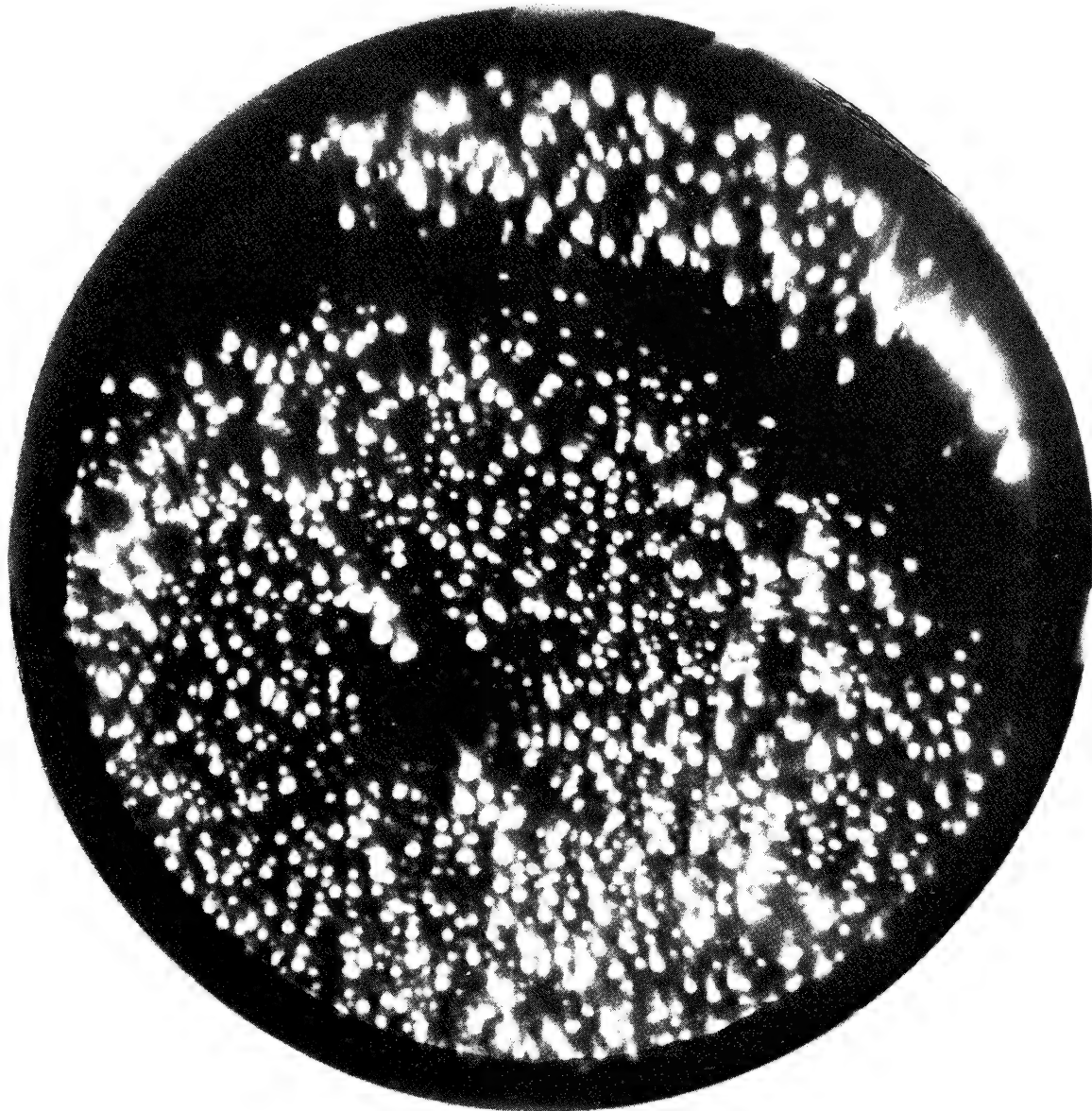


Figure 12. - Tungsten after PTFE contact (17.5 kV, helium image gas, taken after raising voltage to 22.0 kV, liquid helium cooling).



Figure 13. - Tungsten prior to contact (12.0 kV, helium image gas, liquid helium cooling).



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Figure 14. - Tungsten after heavy load PTFE contact.



Figure 15. - Tungsten after heavy load PTFE contact (20.0 kV), helium image gas, after raising voltage to 24.0 kV, liquid helium cooling).

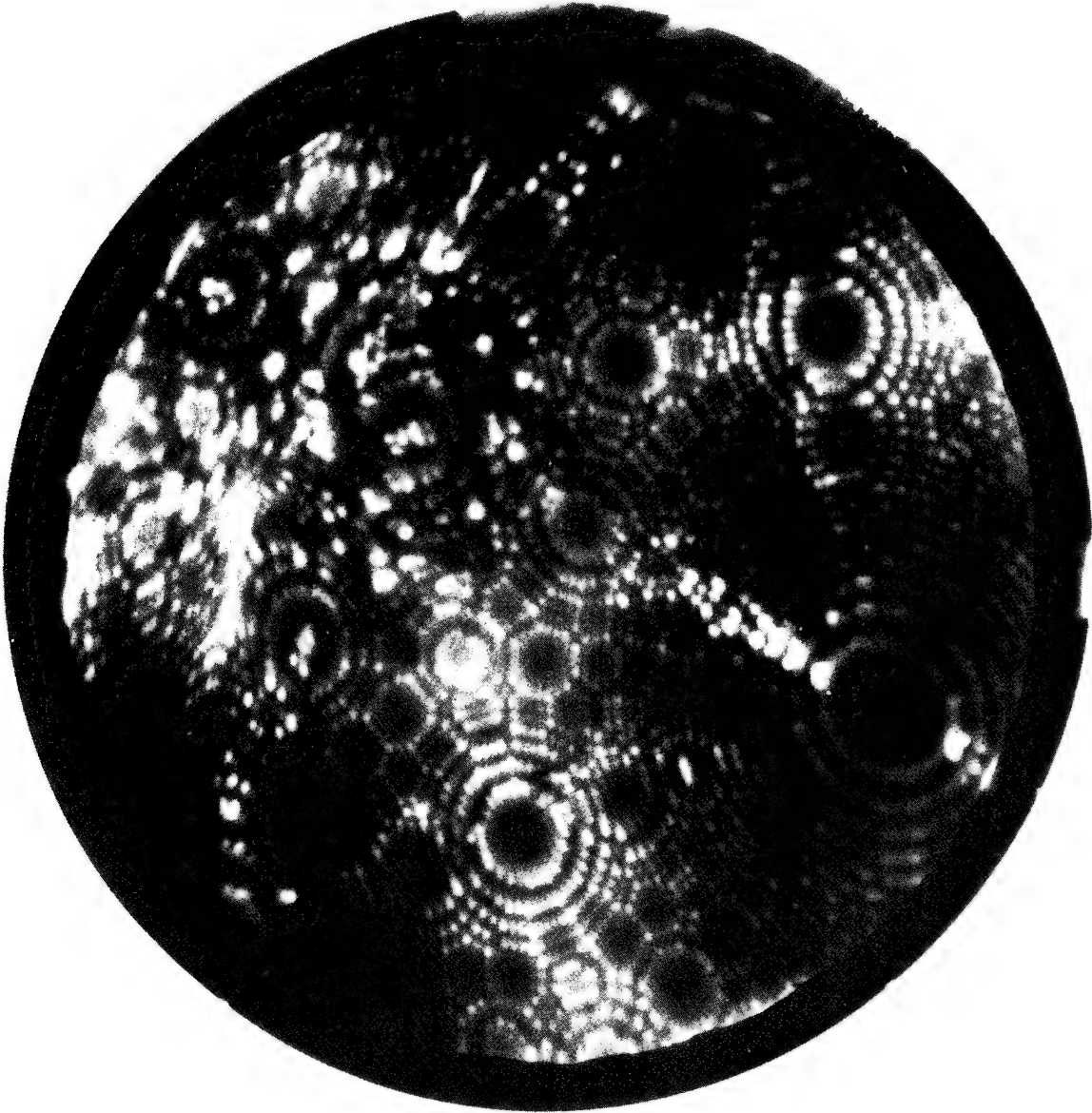


Figure 16. - Tungsten prior to contact (15.0 kV, helium image gas, liquid nitrogen cooling).



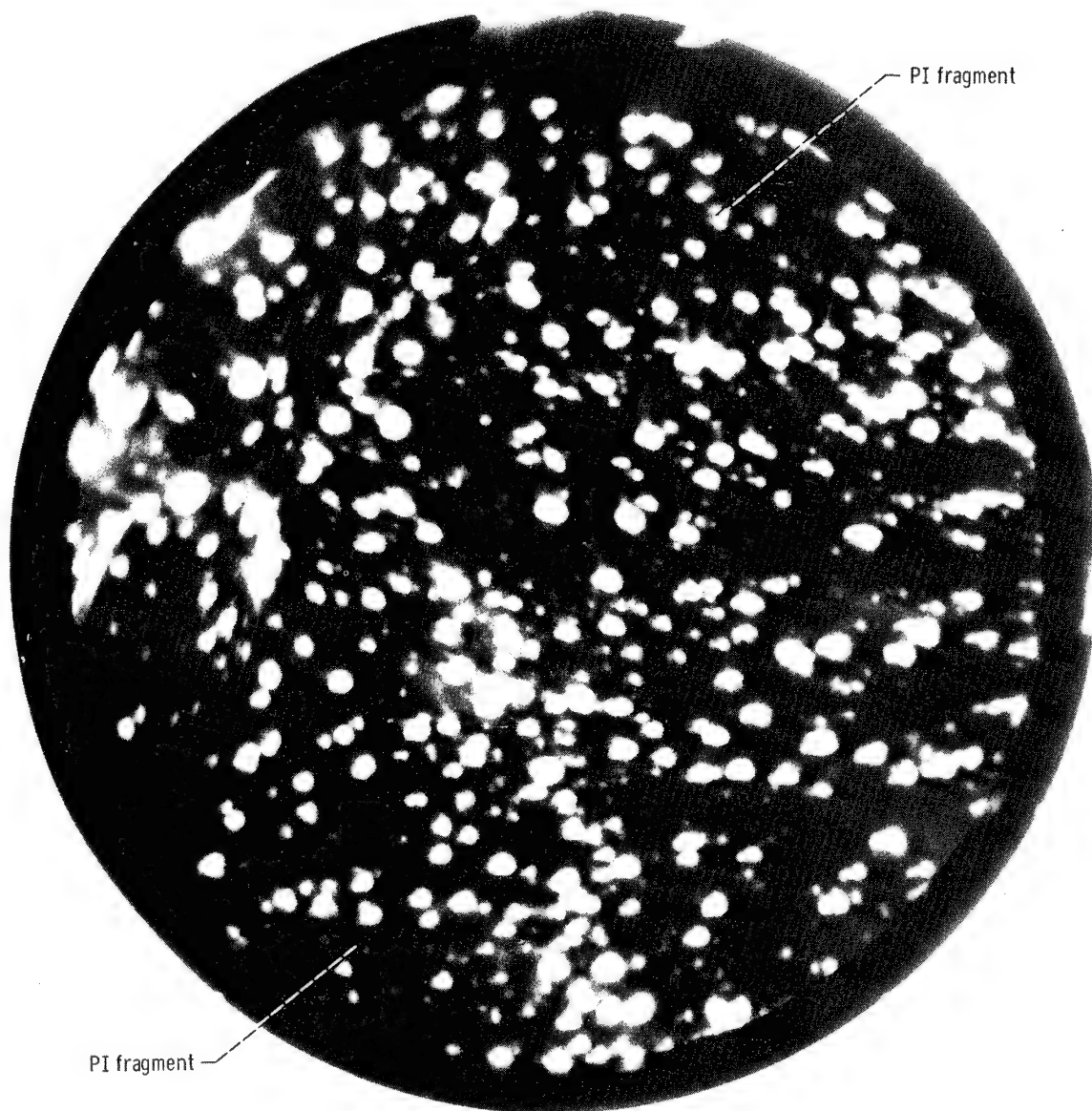


Figure 17. - Tungsten after polyimide contact (13.0 kV, helium image gas, liquid helium cooling).



Figure 18. - Tungsten after polyimide contact (13.5 kV, helium image gas, liquid helium cooling).



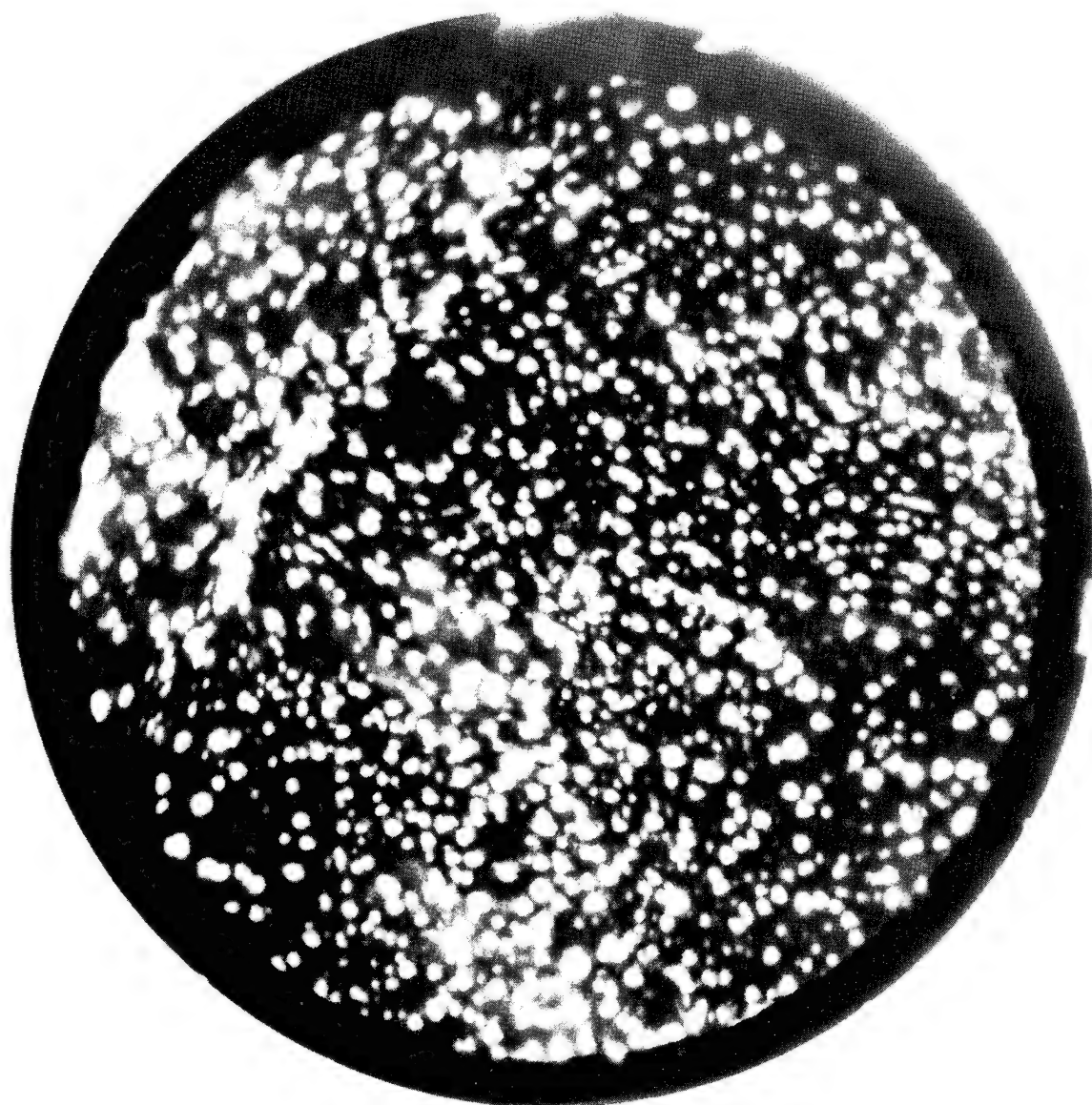


Figure 19. - Tungsten after polyimide contact (14.5 kV, helium image gas, liquid helium cooling).

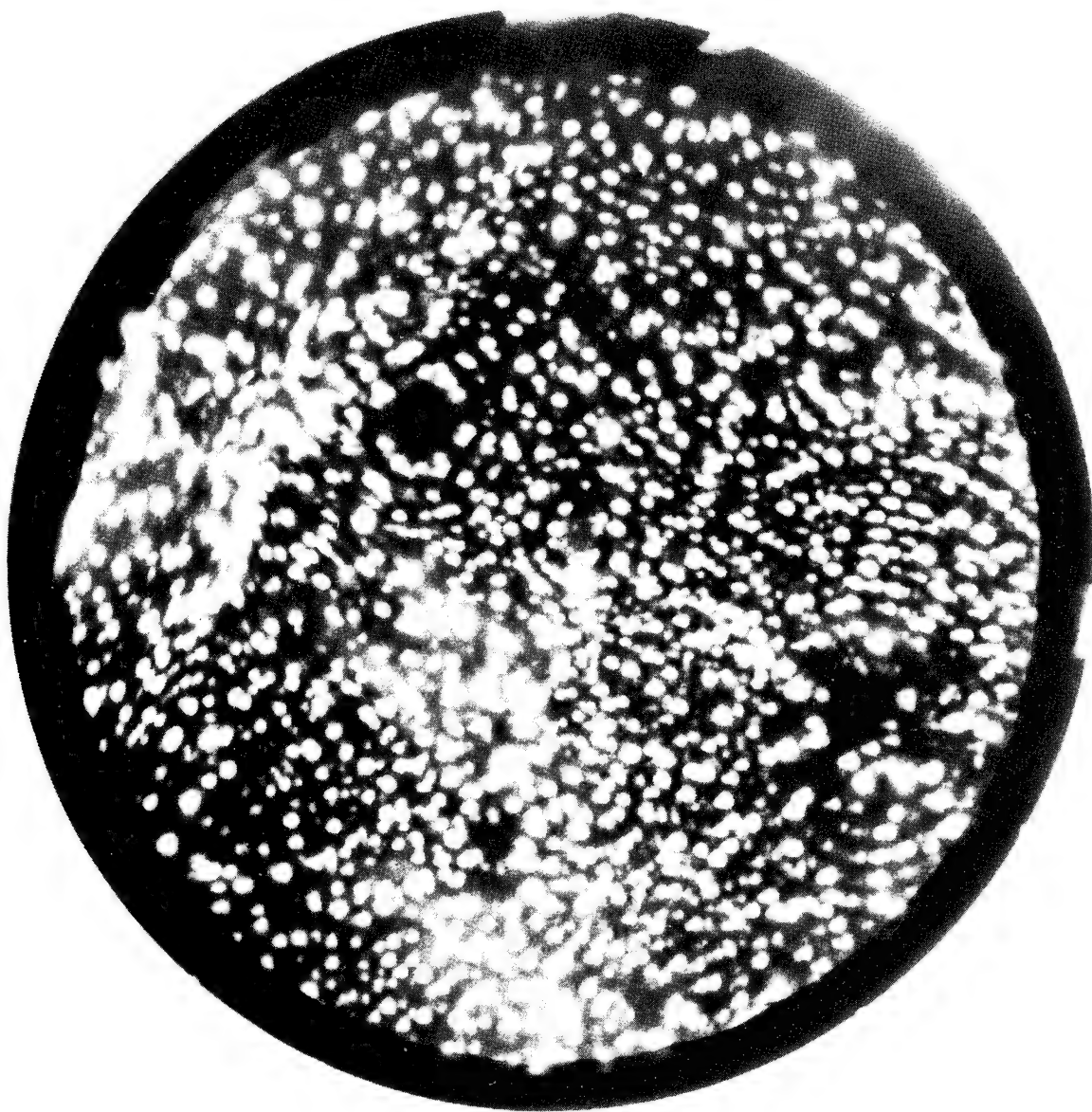


Figure 20. - Tungsten after polyimide contact (15.0 kV, helium image gas, liquid helium cooling).

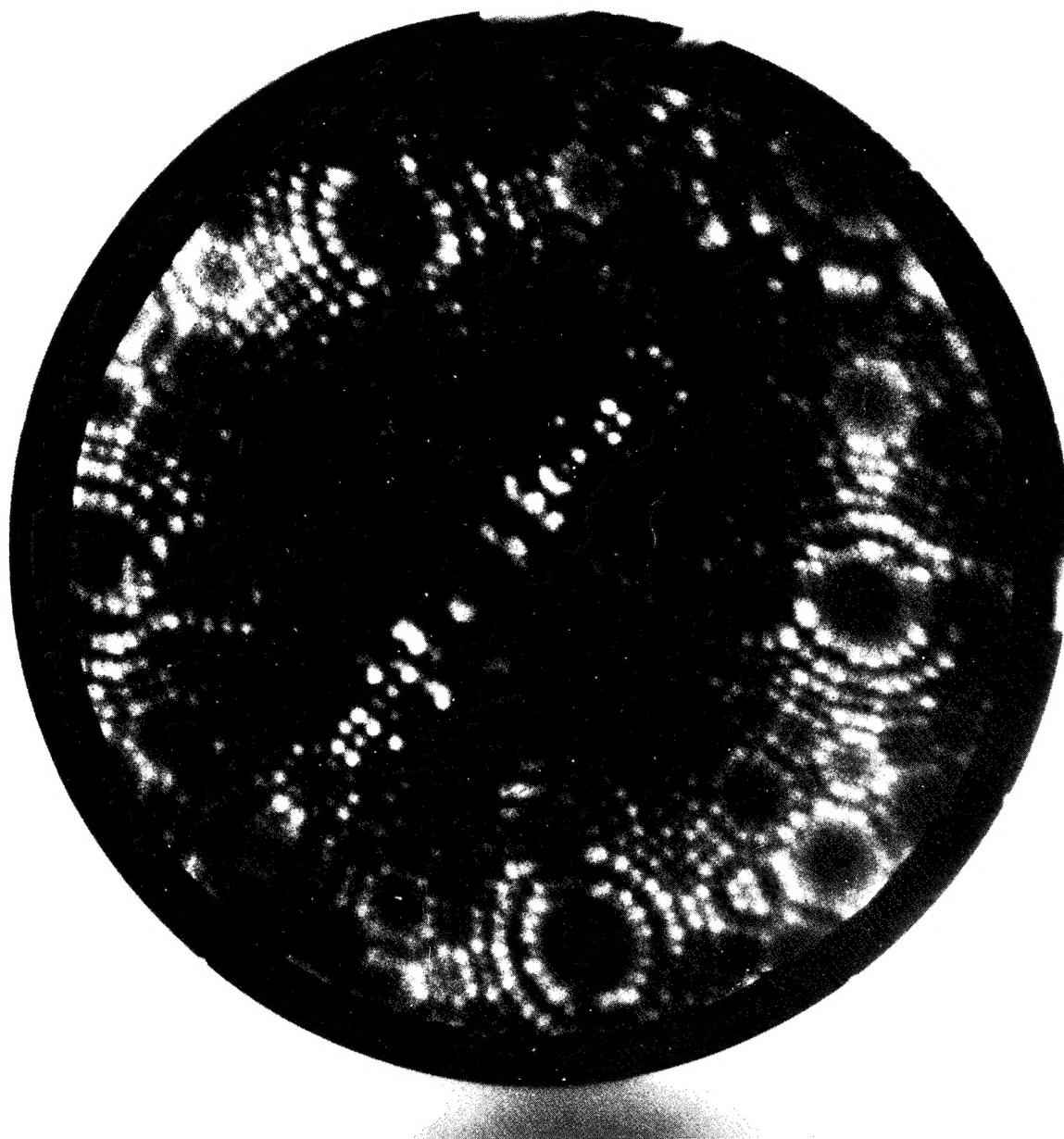
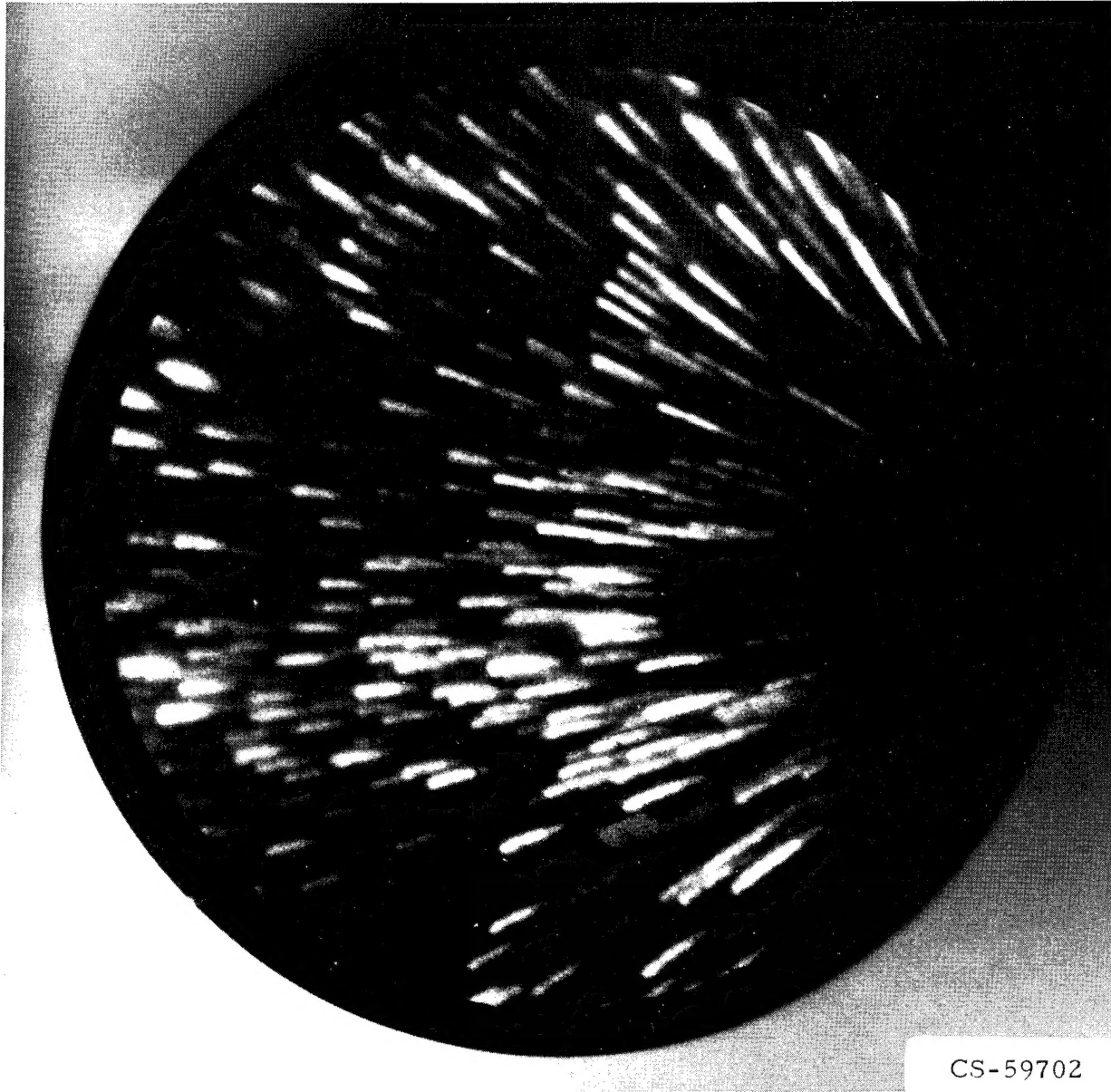


Figure 21. - Tungsten prior to contact (10.5 kV, helium image gas, liquid nitrogen cooling).



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Figure 22. - Tungsten after polyimide contact (9.25 kV, helium image gas, liquid helium cooling).

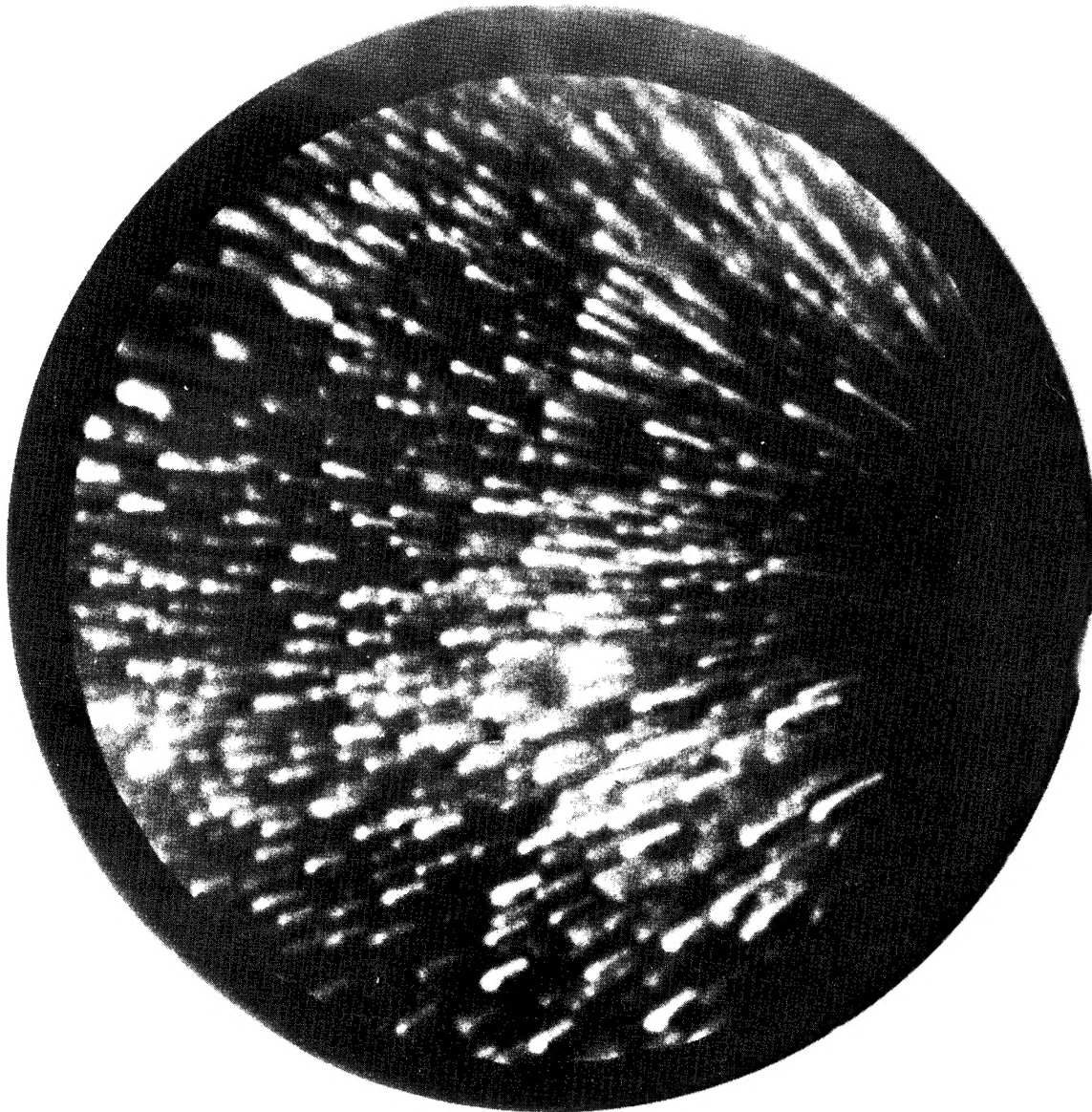


Figure 23. - Tungsten after polyimide contact (9.75 kV, helium image gas, liquid helium cooling).

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